



THE EFFECT OF ALLOYING ELEMENTS ON THE PROPERTIES OF STEELS																							
Alloying element	Mechanical properties						Elasticity	High temperature stability	Cooling rate	Carbide formation	Resistance to wear	Forgeability	Machinability	Scaling	Nitribility	Resistance to corrosion	Magnetic properties						
	Hardness	Strength	Yield Point	Elongation	Reduction of area	Impact value											hysteresis	Permeability	Coercive force	Remanence	Loss of Watt		
Silicon	↑	↑	↑↑	↓	~	↓	↑↑↑	↑	↓	↓	↓↓↓	↓	↓	↓	-	↓	↓	↑↑	↓	↓	-	↓	
Manganese in perlit. steels	↑	↑	↑	~	~	~	↑	~	↓	~	↓	↑	↓	~	~	-							
Manganese in austenit. steels	↓↓↓	↑	↓	↑↑↑	-	-	-	-	↓↓	-	-	↓↓↓	↓↓↓	↓↓	-	-	not magnetic						
Chromium	↑↑	↑↑	↑↑	↓	↓	↓	↑	↑	↓↓↓	↑↑	↑	↓	-	↓↓↓	↑↑	↑↑↑							
Nickel in perlit. steels	↑	↑	↑	~	~	~	-	↑	↓↓	-	↓	↓	↓	-	-								
Nickel in austenit- steels	↓↓	↑	↓	↑↑↑	↑↑	↑↑↑	-	↑↑↑	↓↓	-	-	↓↓↓	↓↓↓	↓↓	-	↑↑	not magnetic						
Aluminium	-	-	-	-	↓	↓	-	-	-	-	-	↓	-	↓↓	↑↑↑	-							
Tungsten	↑	↑	↑	↓	↓	~	-	↑↑↑	↓↓	↑↑	↑↑↑	↓	↓	↓↓	-								
Vanadium	↑	↑	↑	~	~	↓	↑	↑↑	↓↓	↑↑↑	↑↑	↓	↓	↑	↑								
Cobalt	↑	↑	↑	↓	↓	↓	-	↑↑	↑↑	-	↑↑↑	↓	~	↓	-								
Molybdenum	↑	↑	↑	↓	↓	↑	-	↑↑	↓↓	↑↑↑	↑↑	↓	↓	↑↑	↑↑	-							
Copper	↑	↑	↑↑	~	~	~	-	↑	-	-	-	↓↓↓	~	~	-	↑							
Sulphur	-	-	-	↓	↓	↓	-	-	-	-	-	↓↓↓	↑↑↑	-	-	↓							
Phosphorous	↑	↑	↑	↓	↓	↓↓↓	-	-	-	-	-	↓	↑↑	-	-	-							

↑ = Increase ↓=Reduction ~ = constant - = not characteristic or unknown Several arrows = more intensive effect

THE EFFECTS OF ALLOYING ELEMENTS IN STEEL

ELEMENT & SYMBOL	SOLID SOLUBILITY		INFLUENCE UPON FERRITE	INFLUENCE UPON AUSTENITE (HARDENABILITY)	INFLUENCE EXERTED THROUGH CARBIDE		PRINCIPAL FUNCTION OF ELEMENT
	In Gamma Fe	In Alpha Fe			Carbide forming Tendency	Action during Tempering	
Aluminium Al	1.1% (in creased by carbon)	36% ±	Hardens considerably by solid solution.	If dissolved in Austenite increases hardenability mildly.	Graphitizes	-	1. Used as deoxidiser, 2. Restricts grain growth. 3. Alloying element in nitriding steels.
Chromium Cr	12.8% (in 0-5% C steels 20%)-	Unlimited	Hardens slightly, increases corrosion resistance	Increases hardenability moderately. similar to manganese.	Greater than Mn. Less than W.	Mildly resists softening	1. Increases corrosion and oxidation resistance. 2. Increases hardenability. 3. Increases strength at high temperature. 4. With added C resists wear and abrasion.
Cobalt Co	Unlimited	75%	Hardens considerably by solid solution.	Decreases hardenability as dissolved.	Similar to Fe	Sustains hardness by solid solution.	1. Contributes to red hardness by hardening ferrite. 2. Alloying element in high-grade, high-speed steels.
Manganese Mn	Unlimited	3%	Hardens, plasticity somewhat reduced.	Similar to Cr.	Greater than Fe. Less than Cr.	Very little in usual per centages,	1. Counteracts effect of brittleness from sulphur. 2. Increases hardenability inexpensively. 3. High Mn high C produces steels resistant to wear and abrasion,

Continued.

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	In Gamma Fe	In Alpha Fe			Carbide forming Tendency	Action during Tempering	
Molybdenum Mo	3% (in 0.05% C steels 8%)	37.5% (less with lowered temp.)	Age-hardening effect in high Mo-Fe alloys.	Increases hardenability strongly.	Strong. Greater than Cr.	Opposes softening by secondary hardening,	<ol style="list-style-type: none"> 1. Raises coarsening temperature of austenite. 2. Increases depth of hardening. 3. Raises hot and creep strength, red hardness 4. Enhances corrosion resistance in stainless steels. 5. Forms abrasion resistant particules.
Nickel Ni	Unlimited	10% independent of C content,	Strengthens & toughens by solid solution,	Increases hardenability slightly, austenite retention with higher carbon,	Graphitizes less than Fe,	Very little in small percentages.	<ol style="list-style-type: none"> 1. Strengthens unquenched or annealed steels. 2. Toughens pearlitic, ferritic steels (especially at low temperatures) 3. Renders high Cr/Fe alloys austenitic-
Phosphorous P	0.5%	2.8% independent of C content	Hardens strongly by solid solution.	Increases hardenability similar to Mn.	Nil	-	<ol style="list-style-type: none"> 1. Strengthens low C steels. 2. Increases resistance to corrosion. 3. Improves machinability in free cutting steels.
Silicon Si	2%± (in 0.35% C steels 9%)	18.5% (Carbon has little effect)	Hardens with loss in ductility.	Increases hardenability more than Ni	Negative, graphitizes.	Sustains hardness by solid solution.	<ol style="list-style-type: none"> 1. Used as deoxidiser, 2. Alloy for electrical and magnetic sheet steels, 3. Improves oxidation resistance. 4. Strengthens low alloy steels,
Titanium Ti	0.75% (in 0.2% C steels 1%±)	6%± (less with lowered temperatures)	Gives age-hardening in high Fe-Ti alloys.	Probably increases hardenability very strongly as	Greatest known (2% Ti renders 0.5% C steel unhardenable) dissolved. Its carbide effects reduce hardenability.	Some secondary hardening.	<ol style="list-style-type: none"> 1. Fixes carbon in inert particles. 2. Reduces martensitic hardness and hardenability in medium Cr steels. 3. Prevents formation of austenite in high Cr steels. 4. Prevents localised depletion of Cr in stainless steels during long heating periods.
Tungsten W	6% (in 0.25% C steels 11%)	33% (less with lowered temperatures)	Age-hardening system in high W-Fe Alloys.	Increases hardenability strongly in small quantities,	Strong	Opposes softening by secondary hardening.	<ol style="list-style-type: none"> 1. Forms hard, abrasion resistant particles in tool steels, high speed steels. 2. Promotes red hardness and hot strength.
Vanadium V	1.0%± (in 0.2% C steels 4%)	Unlimited	Hardens moderately in solid solution,	Increases hardenability very strongly as dissolved.	Very strong	Maximum for secondary hardening,	<ol style="list-style-type: none"> 1. Promotes fine grain, elevates coarsening temperature of austenite. 2. Increases hardenability when dissolved, 3. Resists tempering and causes marked secondary hardening.

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